

---

## Soil pH

---

### Water (1:1) pH

---

After HACH Company (1992a) and Soil Survey Staff (2014b)

---

#### Application

The 1:1 soil:water is a mixture by weight of one part to one part distilled water. It is the most commonly used method in the field because of the availability of water. Seasonal variations in soil pH can be detected with the 1:1 soil:water method, i.e., it is not used to determine family reaction classes in soil taxonomy (Soil Survey Staff, 2014a). If pH varies widely, knowledge of this variability is important because of the effect of pH on crop performance and on some other aspects of land use. Soil pH is commonly used in conjunction with EC measurements to assess salinity and sodicity. The 1:1 water pH is also a widely used criterion in soil classification (Soil Survey Staff, 2014a).

The 1:2 0.01 *M* CaCl<sub>2</sub> solution is a mixture, by weight, of one part soil to two parts 0.01 *M* CaCl<sub>2</sub> solution. The 0.01 *M* CaCl<sub>2</sub> solution dampens the seasonal variation in soil pH by providing Ca<sup>2+</sup> ions that displace the hydronium and aluminum ions from the colloid surfaces. The result is a pH measurement that remains somewhat invariable to the seasonable changes in pH. Use of the CaCl<sub>2</sub> solution also diminishes the seasonal effect of soluble salt concentration. The CaCl<sub>2</sub> soil pH is generally lower than the 1:1 water pH. The combination of exchange and hydrolysis in salt solutions (0.1 to 1 *M*) can lower the measured pH from 0.5 to 1.5 units, compared to the pH measured in distilled water (Foth and Ellis, 1988). The methods described herein are after the Soil Survey Staff (2014b, methods 4C1a2a1 and 4C1a2a2, respectively) and as applied by HACH Co. (1992a).

#### Summary of Method

An aqueous extract (1:1) is prepared. Contents are stirred for 1 min at 10-min intervals over a 30-min period. The 1:1 pH is measured. The 0.02 *M* CaCl<sub>2</sub> (20 mL) is added to soil suspension, the sample is stirred, and the 1:2 0.01 *M* CaCl<sub>2</sub> pH is measured.

#### Interferences

The difference between the sediment pH and the supernatant pH is called the suspension effect (McLean, 1982). To maintain uniformity in pH determination, measure the pH just above the soil sediment. Clays may cause clogging and slow the electrode response.

Atmospheric CO<sub>2</sub> affects the pH of the soil:water mixture. Closed containers and nonporous materials will not allow equilibration with CO<sub>2</sub>. At the time of pH determination, the partial pressure of CO<sub>2</sub> and the equilibrium point must be considered if critical work is being done. Refer to Appendix 9.3.1 for information about limitations and advantages of pH meters.

### Equipment

1. Scoop
2. Balance (portable)
3. Beakers, polypropylene, 50-mL
4. pH meter, pocket-type or handheld.
5. Gloves, disposable, chemical-resistant (e.g., NSK-24™ Chemical Resistant Nitrile Glove)
6. First-aid kit

### Reagents

1. Distilled water

### Procedure

1. Measure 25 g of soil sample into the 50-mL beaker. Measure 25 mL of distilled water into 25-mL graduated cylinder and transfer into the 50-mL beaker.
2. Stir contents of beaker
3. Wait 1 minute, immerse tip of calibrated pH meter 1 inch (2.5 cm) below the surface of aqueous solution extract and stir gently until soil is completely suspended. Refer to Appendix 9.3.1 on calibration of pH meter.
4. Allow readings to stabilize. Read and record 1:1 soil:water pH.

### Cleanup

5. Rinse electrode with distilled water. Remove excess water by patting it dry with tissue. Allow electrode to dry. Recap and store.

### Answer these questions:

Report the 1:1 soil:water to the nearest 0.1 unit: \_\_\_\_\_

### Use the tables below to answer the following questions:

1. What descriptive term can be used to describe the pH of your soil sample?

*Hint: see Table 1.*

\_\_\_\_\_

2. Is your soil at an ideal pH? If not, why? *Hint: see table 2*

\_\_\_\_\_  
\_\_\_\_\_  
\_\_\_\_\_

3. Is your soil a suitable pH for growing the following crops? *Hint: use table 3*

Corn? \_\_\_\_\_

Lettuce? \_\_\_\_\_

Potatos (Sweet)? \_\_\_\_\_

**Table 1.—Descriptive Terms Commonly Associated With Certain 1:1 pH Ranges (Soil Survey Division Staff, 1993).**

Term	pH
Extremely acid	<4.5
Very strongly acid	4.5–5.0
Strongly acid	5.1–5.5
Moderately acid	5.6–6.0
Slightly acid	6.1–6.5
Neutral	6.6–7.3
Slightly alkaline	7.4–7.8
Moderately alkaline	7.9–8.4
Strongly alkaline	8.5–9.0
Very strongly alkaline	>9.1

**Table 2.—Agronomic Interpretations (Indications and Associated Conditions) of pH Ranges (HACH Co, 1993; Ryan et al., 2001).**

pH	Interpretation
pH <5.5	Soil is deficient in Ca and Mg and should be limed. Poor root growth due to low cation-exchange capacity (CEC) and possible Al <sup>3+</sup> toxicity. Phosphorus deficiency is likely.
pH 5.5–6.5	Soil is low in carbonate but should be monitored. Satisfactory for many crops.
pH 6.5–7.5	Ideal range for most crops. Soil CEC is near 100%.
pH 7.5–8.4	Free carbonate present in soil. Usually excellent infiltration and percolation of water related to high Ca saturation of clays. Typically P and micronutrients less available.
pH >8.4	Typically, indicative of sodic soil. Poor soil physical conditions. Low infiltration and percolation. Possible root deterioration and organic matter dissolution.

**Table 3.—Suitable Soil pH (1:1) Ranges for Selected Crops (after Whittaker et al., 1959).**

Crops	Soil pH Ranges									
	4.5	5.0	5.5	6.0	6.5	7.0	7.5			
Alfalfa										
Alsike clover										
Apples										
Asparagus										
Barley										
Beans, lima										
Beans, snap										
Beans, velvet										
Blueberries										
Buckwheat										
Cabbage										
Carrots										
Clover, crimson										
Clover, red										
Clover, sweet										
Clover, white										
Corn										
Cotton										
Cowpeas										
Cucumber										
Grasses										
Kale										
Lettuce										
Mustard										
Oats										
Onions										
Parsnips										
Peas										
Peppers										
Potatoes, sweet										
Potatoes, white										
Radishes										
Rye										
Sorghum										
Soybeans										
Spinach										
Squash										
Strawberries										
Sudan grass										
Timothy										
Tobacco										
Tomatoes										
Trefoil, birdsfoot										
Vetch										
Wheat										10